## Direct catalytic asymmetric anti-selective Mannich-type reactions†

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The simple chiral pyrrolidine catalyzed asymmetric antiselective Mannich-type reaction is presented; the reaction gives the corresponding amino acid derivatives with 10:1->19:1 dr and 97-99% ee.

The development of direct catalytic Mannich reactions has received considered attention in recent years. It is used for the enantioselective synthesis of amino acids, β-lactams, amino sugars, imino sugars and amino alcohols. 1-12 The synthetic utility of these products have created a demand for catalytic highly enantioselective Mannich reactions that are either syn- or anti-selective.<sup>2</sup> Several catalysts have been developed for the catalysis of synselective direct asymmetric Mannich reactions. For instance, organometallic La-,3 Zn-,4 Cu-5 and In-complexes6 catalyze the reaction with excellent syn- and enantioselectivity. Moreover, organocatalysts such as Brønsted acids,7 chincona alkaloids,8 proline and its derivatives,<sup>9</sup> proline tetrazoles,<sup>10</sup> and amino acids catalyze<sup>11</sup> the direct Mannich reaction with high syn- and enantioselectivity. However, there are only a few examples of anti-selective direct catalytic asymmetric Mannich-type reactions. 12 Thus, the development of direct catalytic asymmetric anti-selective Mannich reactions is important and challenging.<sup>13</sup> Herein, we report a highly anti- and enantioselective Mannich-type reaction with aldehydes as nucleophiles that is catalyzed by a simple diphenylprolinol derivative.

In the proline, hydroxyproline and proline tetrazole-catalyzed Manich reactions, the Si-facial attack on the imine with a trans configuration by the Si-face of the chiral enamine gives the corresponding Mannich product with a syn relative configuration (Fig. 1). 9,10 However, we believed that efficient shielding of the Siface of the chiral enamine by the employment of a bulky chiral amine catalyst would switch the facial selectivity of the reaction

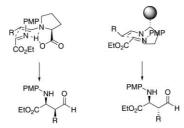


Fig. 1 The stereochemical outcome of the (S)-proline direct asymmetric catalyzed Mannich reaction and a plausible bulky chiral amine catalyzed Mannich-type reaction.

Department of Organic Chemistry, Arrhenius Laboratory, Stockholm University, Sweden. E-mail: acordovala@netscape.net; acordova@organ.su.se; Fax: + 46 8 15 49 08; Tel: +46 8 162479 † Electronic supplementary information (ESI) available: experimental procedures. See DOI: 10.1039/b602221a

(Fig. 1) and thus change the stereochemical outcome of the reaction and make it anti-selective. To test this we investigated the chiral pyrrolidine-catalyzed Mannich-type reaction between ethyl N-p-methoxyphenyl (PMP)-protected α-iminoglyoxylate and isovaleraldehyde 2a (eqn (1)). We found that several bulky pyrrolidine derivatives catalyzed the reaction with moderate to good antiselectivity. To our delight we found that the readily available TMS-protected α,α-diphenyl-2-pyrrolidinemethanol<sup>14</sup> (diphenylprolinol, 1) catalyzed the direct catalytic asymmetric Mannichtype reaction with high anti- and enantioselectivity and the corresponding β-formyl-α-amino acid 3a was furnished in 56% yield with >19:1 dr (anti: syn) and 92% ee.

Encouraged by this result, we performed a solvent screen with protected diphenylprolinol 1 as the catalyst (Table 1).

The reaction proceeded smoothly with high anti- and enantioselectivity in all the solvents tested. The highest enantioselectivity was obtained in CHCl<sub>3</sub> and CH<sub>3</sub>CN. Notably, the reaction was highly stereoselective in water: the amino acid derivative 3a was formed with 12:1 dr and 98% ee. In fact, this is the first example of a direct catalytic asymmetric Mannich-type reaction in water. Based on the results from the solvent screen we chose to investigate the chiral pyrrolidine 1 catalyzed anti-selective reaction between different aldehydes and N-PMP-protected α-iminoglyoxylate in CHCl<sub>3</sub> and CH<sub>3</sub>CN (Table 2).

Table 1 Solvent screen for the direct catalytic asymmetric antiselective formation of 2a

EtO <sub>2</sub> C H + 2a		+ OH 2a		1 10 mol%) Solvent	PMP NH O EtO <sub>2</sub> C H		
Entry	Solvent		Time (h)	Temp. (°C)	Yield (%) <sup>a</sup>	$dr^b$	ee (%) <sup>c</sup>
1	t-BuOH:	H <sub>2</sub> O-1·1	16	rt	35	>19:1	96
2	CH <sub>3</sub> CN	1120 1.1	17	rt	50	>19:1	99
3	CHCl <sub>3</sub>		16	rt	56	>19:1	92
4	CHCl <sub>3</sub>		24	4	68	>19:1	98
5	Toluene		16	rt	43	14:1	88
6	EtOH		16	rt	32	>19:1	82
7	H <sub>2</sub> O		16	4	35	12:1	98

<sup>a</sup> Isolated yield of the pure products after silica-gel chromatography. Anti:syn ratio as determined by NMR analyses. <sup>c</sup> Determined by chiral-phase HPLC analyses. PMP = p-methoxyphenyl.

**Table 2** Direct catalytic *anti*-selective asymmetric Mannich-type reactions

Entry	R	Cond.	Prod		Yield (%) <sup>a</sup>	$dr^b$	ee (%) <sup>c</sup>
1	<i>i</i> -Pr	A	3a	24	68	>19:1	98
2	<i>n</i> -pent	A	3b	18	63	>19:1	99
3	Me	A	3c	14	$45 (35)^d$	$14:1(15:1)^d$	99 $(90)^d$
4	Me	В	3c	17	75	15:1	99
5	$CH_2OBn$	A	3d	16	$67 (30)^d$	$14:1(10:1)^d$	$97 (90)^d$
6	$CH_2Ph$	A	3e	16	67	19:1	99
7	i-Pr	A	3a	16	45 <sup>e</sup>	$>19:1^{e}$	99 <sup>e</sup>

<sup>a</sup> Isolated yield of the pure products after silica-gel chromatography. <sup>b</sup> Syn:anti ratio as determined by NMR analyses. <sup>c</sup> Determined by chiral-phase HPLC analyses. d Reaction performed in H2O at 4 °C, 1 h reaction time. Bn = benzyl, PMP = p-methoxyphenyl. <sup>e</sup> TMS protected di(2-naphthyl)prolinol (10 mol%) was used as the catalyst.  $A = CHCl_3$ , 4 °C.  $B = CH_3CN$ , 4 °C.

The diphenylprolinol 1-catalyzed direct Mannich-type reactions were highly enantioselective and amino acid derivatives 3 were isolated in moderate to high yields with 97-99% ee. The reactions were highly *anti*-selective (14 : 1 $\rightarrow$ 19 : 1). For instance,  $\alpha$ -amino acid 3e was furnished in 67% yield with 19:1 dr and 99% ee. Moreover, the chiral amine-catalyzed asymmetric reaction with α-benzyloxyacetaldehyde as the donor was highly stereoselective and gave 3d in 67% yield with 14:1 dr and 97% ee. Amino sugar 3d with an anti-configuration is an important chiral synthon and can be used in the de novo synthesis of C-6 amino- and iminosugars. 91 Thus, the catalytic asymmetric anti-selective Mannich-type reaction opens up the possibility of synthesizing all the different diastereomers of amino and imino sugar derivatives. We also investigated the organocatalytic asymmetric Mannich-type reactions in water with aldehydes 2 as nucleophiles. The reactions were very fast due to the hydrophobic effect<sup>15</sup> and the product together with the starting imine formed an organic precipitate within 1 h, which upon isolation gave the corresponding products 3 with high enantioselectivity (90–98% ee) but low conversion (<40%). Moreover, TMS protected di(2-naphthyl)prolinol catalyzed the asymmetric Mannich reaction with excellent anti- and enantioselectivity to give 3a with >19:1 dr and 99% ee.

The absolute stereochemistry of the Mannich products was established by comparison with the epimerized (2S,3S)-syn-3a and **3d** Mannich products obtained by (S)-proline catalysis.  ${}^{9g,i,l}$  Hence, (S)-diphenylprolinol 1 catalyzed the asymmetric formation of (2S,3R)-amino acid derivatives 3. The stereochemical outcome of the reaction was explained by the proposed transition state I (Fig. 2). Thus, attack on the Si-face of the imine with a transconfiguration by the Re-face of the chiral enamine gives the amino acid derivative 3.

Stabilization of the trans-configuration and efficient shielding of the Si-face of the chiral enamine can explain the high stereoselectivity of the reaction. In addition, plausible stabilization by coulombic interactions between the amine atom of the imine and the  $\delta^+$  on the nitrogen of the pyrrolidine moiety of the chiral enamine, which is generated during the nucleophilic attack,

TMSO 
$$\stackrel{Ar}{=}$$
 Ar  $\stackrel{R}{=}$   $\stackrel{N}{=}$  Ar  $\stackrel{R}{=}$   $\stackrel{N}{=}$   $\stackrel{Ar}{=}$   $\stackrel{R}{=}$   $\stackrel{N}{=}$   $\stackrel$ 

Fig. 2 Proposed transition state for the protected diarylprolinol catalyzed direct asymmetric anti-selective Mannich-type reaction.

contributes to the stabilization of the Si-facial attack on the electrophile. 16

In summary, we have developed a highly anti- and enantioselective direct catalytic asymmetric Mannich-type reaction. The asymmetric Mannich-type reactions are catalyzed by simple diphenyl- and di(2-naphthyl)prolinol derivatives that are prepared in one step and furnish amino acid derivatives in high yields with 14: 1->19: 1 dr and up to 99% ee. The reaction is also highly stereoselective in water (90-98% ee). Further studies on the development of direct catalytic anti-selective Mannich reactions and their application in catalytic asymmetric domino- and tandemreactions are ongoing.

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## Notes and references

- 1 A. Córdova, Acc. Chem. Res., 2004, 37, 102.
- 2 (a) S. Kobayashi, H. Ishitani and M. Ueno, J. Am. Chem. Soc., 1998, **120**, 431; (b) T. Hamada, K. Manabe and S. Kobayashi, J. Am. Chem. Soc., 2004, 126, 7768; (c) S. Kobayashi and H. Ishitani, Chem. Rev., 1999, 99, 1069; (d) S. Kobayashi, R. Matsubara, Y. Nakamura, H. Kitagawa and M. Sugiura, J. Am. Chem. Soc., 2003, 125, 2507.
- 3 S. Yamasaki, T. Iida and M. Shibasaki, Tetrahedron Lett., 1999, 40, 307.
- 4 (a) S. Matsunaga, N. Kumagai, N. Harada, S. Harada and M. Shibasaki, J. Am. Chem. Soc., 2003, 125, 4712; (b) B. M. Trost and L. M. A. Terrell, J. Am. Chem. Soc., 2003, 125, 338.
- 5 K. Juhl, N. Gathergood and K. A. Jørgensen, Angew. Chem., Int. Ed., 2001, 40, 2995.
- 6 S. Harada, S. Handa, S. Matsunaga and M. Shibasaki, Angew. Chem., Int. Ed., 2005, 43, 4365
- 7 T. Akiyama, J. Itoh, K. Yokota and K. Fuchibe, Angew. Chem., Int. Ed., 2004, 43, 1566; D. Uraguchi and M. Terada, J. Am. Chem. Soc., 2004. 126. 5356.
- 8 S. Lou, B. M. Taoka, A. Ting and S. Schaus, J. Am. Chem. Soc., 2005, **127**. 11256.
- 9 (a) B. List, J. Am. Chem. Soc., 2000, 122, 9336; (b) A. Córdova, Chem.-Eur. J., 2004, 10, 1987; (c) A. Córdova, Synlett, 2003, 1651; (d) Y. Hayashi, W. Tsuboi, I. Ashimine, T. Urushima, M. Shoji and K. Sakai, Angew. Chem., Int. Ed., 2003, 42, 3677; (e) A. Münch, B. Wendt and M. Christmann, Synlett, 2004, 2751; (f) W. Zhuang, S. Saaby and K. A. Jørgensen, Angew. Chem., Int. Ed., 2004, 43, 4476; (g) A. Córdova, S.-I. Watanabe, F. Tanaka, W. Notz and C. F. Barbas, III, J. Am. Chem. Soc., 2002, 124, 1866; (h) S. Fustero, D. Jimenez, J. F. Sanz-Cervera, M. Sanchez-Rosello, E. Esteban and A. Simon-Fuentes, Org. Lett., 2005, 7, 3433; (i) A. Córdova and C. F. Barbas, III, Tetrahedron Lett., 2002, 43, 7749; (j) B. Westermann and C. Neuhaus, Angew. Chem., Int. Ed., 2005, 44, 4077; (k) I. Ibrahem and A. Córdova, Tetrahedron Lett., 2005, 46, 2839; (I) W.-W. Liao, I. Ibrahem and A. Córdova, Chem. Commun., 2006, 674; (m) D. Enders, C. Grondal, M. Vrettou and G. Raabe, Angew. Chem., Int. Ed., 2005, 44, 4079.
- 10 A. J. A. Cobb, D. M. Shaw and S. V. Ley, Synlett, 2004, 558; A. J. A. Cobb, D. M. Shaw, D. A. Longbottom, J. B. Gold and S. V. Ley, Org. Biomol. Chem., 2005, 3, 84.
- I. Ibrahem, W. Zou, M. Engqvist and Y. Xu, Chem.-Eur. J., 2005, 11,
- (a) T. Yoshida, H. Morimoto, N. Kumagai, S. Matsunaga and M. Shibasaki, Angew. Chem., Int. Ed., 2005, 44, 3470; (b) T. Kano,

- Y. Yamaguchi, O. Tokuda and K. Maruoka, *J. Am. Chem. Soc.*, 2005, **127**, 16408; (*c*) S. Mitsumori, H. Zhang, P. H.-Y. Cheong, K. N. Houk, F. Tanaka and C. F. Barbas, III, *J. Am. Chem. Soc.*, 2006, **128**, 1040; (*d*) B. M. Trost, J. Jaratjaroonphong and U. Reutrakul, *J. Am. Chem. Soc.*, 2006, **128**, 2778.
- 13 M. M. B. Marques, Angew. Chem., Int. Ed., 2006, 45, 348.
- 14 (a) E. J. Corey, R. K. Bakshi and S. Shibata, J. Am. Chem. Soc., 1987, 109, 5551; (b) E. J. Corey, Angew. Chem., Int. Ed., 2002, 41, 1650. Both enantiomers of α,α-Diphenylprolinol and α,α-di(2-naphthyl)prolinol are commercially available and TMS protected in one-step see; (c) M. Marigo, T. C. Wabnitz, D. Fielenbach and K. A. Jørgensen, Angew. Chem., Int. Ed., 2005, 44, 794; (d) M. Marigo, D. Fielenbach,
- A. Braunton, A. Kjaersgaard and K. A. Jørgensen, Angew. Chem., Int. Ed., 2005, 44, 3703; (e) J. Franzén, M. Marigo, D. Fielenbach, T. C. Wabnitz, A. Kjaersgaard and K. A. Jørgensen, J. Am. Chem. Soc., 2005, 127, 18296; (f) M. Marigo, T. Schulte, J. Franzén and K. A. Jørgensen, J. Am. Chem. Soc., 2005, 127, 15710; (g) Y. Hayashi, H. Gotoh, T. Hayashi and M. Shoji, Angew. Chem., Int. Ed., 2005, 44, 4212; (h) H. Sundén, I. Ibrahem and A. Córdova, Tetrahedron Lett., 2006, 47, 99.
- (a) R. Breslow, Acc. Chem. Res., 1991, 21, 159; (b) S. Narayan,
   J. Muldoon, M. G. Finn, V. V. Fokin, H. C. Kolb and K. B. Sharpless,
   Angew. Chem., Int. Ed., 2005, 44, 3275.
- 16 V. Vinkovic and V. Sunjic, Tetrahedron, 1997, 53, 689.

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